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## **Through-Vial Impedance Spectroscopy (TVIS)**

A novel process analytical technology for the development of pharmaceutical products and processes



## 2nd Annual Pharmaceutical Lyophilisation Summit



#pharmalyo18

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- Description of TVIS measurement system
- Applications in Brief
- First time report on the use of dual-electrode system and its applications
  - > Ice region specific temperature prediction  $(T_i, T_b)$
  - Drying rate determination
  - > Heat transfer coefficient ( $K_v$ ) determination
- Acknowledgements
- TVIS dielectric loss mechanisms







# **Through Vial Impedance Spectroscopy (TVIS)** Description of Measurement System



## Introduction to the TVIS System

- Impedance spectroscopy characterizes the ability of materials to conduct electricity under an applied an oscillating voltage (of varying frequency)
- Impedance measurements across a vial rather than within the vial
- Hence "Through Vial Impedance Spectroscopy"
- Features
  - Single vial "non-product invasive"
  - Both freezing and drying characterised in a single technique
  - Non-perturbing to the packing of vials
  - Stopper mechanism unaffected











## Freeze drying chamber



LyoView<sup>™</sup> analysis software



LyoDEA<sup>™</sup> measurement software

**Resultant** Stimulating

*current* 







**TVIS system** 

(I to V convertor)





## Through Vial Impedance Spectroscopy (TVIS) Applications



## **Through Vial Impedance Spectroscopy (TVIS)**



*C*'(~ 100 kHz) is highly sensitive to low ice volumes; therefore it could be used for determination **end point** of primary drying



## **TVIS Response Surface (3D-Plot)**







## **Dielectric loss spectrum**



 Data analysing software (LyoView ™) identifies the peak frequency (F<sub>PEAK</sub>) and peak amplitude (C"<sub>PEAK</sub>) in the imaginary part of the capacitance spectrum







# Through Vial Impedance Spectroscopy (TVIS)

## Dual-electrode system and its applications

(Ice temperature, Drying rate and Heat transfer coefficient)







- A dual electrode system comprises two pairs of copper electrode glued to the external surface of a Type I tubular glass vial.
- This option is suitable for large volume samples, including those used for  $K_v$  determination.



## **Temperature Determination**





•  $T_{(F_{PEAK})TE}$  : TVIS predicted temperature from top electrode (TE)

•  $T_{(F_{PEAK})BE}$  : TVIS predicted temperature

from bottom electrode (BE)

Both  $T_i$  and  $T_b$  can be estimated by extrapolating from the temperatures predicted from the centers of top electrode  $(T_{(F_{PEAK})TE})$  and bottom electrode ( $T_{(F_{PEAK})BE}$ ).







## Objective

Temperature calibration of log  $F_{PEAK}$  of top electrode  $(T_{(F_{PEAK})TE})$  and bottom electrode  $(T_{(F_{PEAK})BE})$ 

Annealing the sample

In-line TVIS measurement

Ι

Identifying peak frequency (F<sub>PEAK</sub>) using LyoView ™ software

Calibration plot (temperature vs Log F<sub>PEAK</sub>)

Predicting product temperature using calibration plot









The product temperature predicted by TVIS can demonstrate the temperature gradient across ice cylinder height









# **Objective** IV Compensation of $C''_{PEAK}$ during primary drying

## Compensation of $C''_{PEAK}$ during primary drying

• During primary drying,  $C_{PEAK}''$  is attributed to both the loss of ice and product temperature; therefore, it requires a standardization factor ( $\emptyset$ ) for temperature compensation:

**Objective** 

$$\phi(T) = \frac{C_{PEAK}''(T)}{C_{PEAK}''(T_{ref})}$$

 $C_{PEAK}''(T)$  and  $C_{PEAK}''(T_{ref})$  are the peak amplitudes at temperatures (T) and reference temperature ( $T_{ref}$ ) during the re-heating ramp. In this presentation, a temperature of -20 °C is used as the reference temperature value

• The expression for  $\emptyset(T)$  can be re-written in terms of the polynomial coefficients (slide 22):

$$\phi(T) = \frac{aT^2 + bT + c}{aT_{ref}^2 + bT_{ref} + c}$$

• Values of  $C''_{PEAK}$  during primary drying are then standardized to the reference temperature by dividing by  $\emptyset(T)$  to give a standardized peak amplitude of  $\hat{C}''_{PEAK}$ 

$$\hat{\mathsf{C}}_{PEAK}'' = \frac{\mathcal{C}_{PEAK}''(T)}{\emptyset(T)}$$



The standardized  $C''_{PEAK}$  is defined as  $\hat{C}''_{PEAK}$ 





#### **Objective** Calibration of $C''_{PEAK}$ for ice layer height 2.0 -C" (Imag. part / pF) 15 mm 1.5 1.0 0 mm 0.5 0 mm 0 mm 0 mm 0.0 Freeze drying shelf Freeze drying shelf Freeze drying shelf 3 4 Log Frequency 1 2 5 6 Freezing water at -20°C Thawing ice at 5°C for Filling water into TVIS vial up to bottom for 3 hours 3 hours edge of top electrode Û (≡ 0 mm) Repeat this process Е<sup>16</sup> Е14 until the ice height reaches top edge of Ice height from bottom top electrode (15 mm) electrode/ 9 & 0 2 ~0.92 mm ~13.8 mm Freeze drying shelf of An aliquot part (0.35 g) 4 Freeze drying shelf of water corresponding edge 0 to ice height 1 mm is added into TVIS vial 0 0.5 0.0 1.0 1.5

23

 $-C''_{PEAK} / pF$ 



**Objective** VI Estimation of ice layer height during primary drying



**Objective** VI Estimation of ice layer height during primary drying

- The dependency of  $C_{PEAK}''$  on the ice cylinder height in linear region
- Surrogate drying rate can be estimated in terms of decreasing ice height











The product temperature at ice interface predicted by using a 2-points temperature extrapolation close to the temperature of ice vapour at chamber pressure of 270 µbar  $(T_{(P_i=P_{C@270\mu bar})})$ 



# **Objective** VIII Comparison of TVIS drying rate ( $\Delta m/\Delta t$ ) with gravimetric method (weight loss)





- Drying rate is based on the assumption of a planar sublimation front
- The change in ice cylinder height (h) can be equated to the change in ice volume (v)

$$v(cylinder) = \pi r^2 h = Ah$$

Where r is internal radius of vial and A is internal cross section area of vial (=  $\pi r^2$ )

• Ice volume can be converted to ice mass (m) by multiplying with ice density ( $\rho_i$ )

$$m = \rho_i \cdot \pi r^2 h = \rho_i \cdot A h$$

• Hence; drying rate  $\left(\frac{\Delta m}{\Delta t}\right)$  can be expressed by

Drying rate 
$$(\frac{\Delta m}{\Delta t}) = \rho_i \cdot A \cdot \frac{h_{(t1)} - h_{(t2)}}{t_2 - t_1}$$





• An average surrogate drying rate calculation

Drying rate 
$$(\frac{\Delta m}{\Delta t}) = \rho_i \cdot A \cdot \frac{h_{(t1)} - h_{(t2)}}{t_2 - t_1}$$

Ice density ( $\rho_i$ ) at -32 °C Internal vial diameter (VC010-20C) Cross-section area (A) Ice height at 0 h ( $h_{(0 h)}$ ) Ice height at 4.9 h ( $h_{(4.9 h)}$ )



Drying rate = 0.920 
$$g \cdot cm^{-3} \times 3.80 \ cm^2 \times \frac{(20.50 - 16.68) \times 10^{-1} cm^2}{(4.9 - 0)h}$$

 $= 0.920 \text{ g} \cdot \text{cm}^{-3}$ 

= 2.21 cm

 $= 3.80 \text{ cm}^2$ 

= 20.50 mm

= 16.68 mm

$$= 0.27 \ g \cdot h^{-1}$$

	Drying rate		
TVIS	0.27 g/h		
Gravimetric	0.25 g/h		



# **Objective** IX Determination (i) the drying rate $(\Delta m / \Delta t)$ and (ii) ice base temperature $(T_b)$ during the steady state period

Determination (i) the drying rate  $(\Delta m/\Delta t)$  and (ii) ice base temperature  $(T_b)$  during the steady state period for heat transfer coefficient  $(K_v)$  calculation



• Drying rate during the steady state

Drying rate 
$$(\frac{\Delta m}{\Delta t}) = \rho_i \cdot A \cdot \frac{h_{(t1)} - h_{(t2)}}{t_2 - t_1}$$

Ice density  $(\rho_i)$  at -32°C= 0.920 g·cm<sup>-3</sup>(Calculated ice temperature between  $T_i \& T_b$ )Internal vial diameter (VC010-20C)= 2.21 cmCross-section area (A)= 3.80 cm<sup>2</sup>Ice height at 2 h  $(h_{(2 h)})$ = 19.94 mmIce height at 2.8 h  $(h_{(2.8 h)})$ = 18.98 mm

TVIS parameters used for determination:  $\frac{\Delta m}{\Delta t} = 0.42 \text{ g} \cdot \text{h}^{-1}$   $T_{h} = -29.8^{\circ}\text{C}$ 

Drying rate = 0.920 
$$g \cdot cm^{-3} \times 3.80 \ cm^2 \times \frac{(19.94 - 18.98) \times 10^{-1} cm}{(2.8 - 2.0) \ h}$$
  
= 0.42  $g \cdot h^{-1}$ 



**Objective** 

-28

-32

-36

-40

-44

21

20

19

18

17

16

0

Ice height (h)/mm

0

1

 $h_{(2.8 h)}$  18.98 mm

1

Temperature /°C

IX

źime / h

2 Time / h

 $T_{avg} \sim 32 \degree C \sim T_b$ 

Temp. constant

 $\frac{T_i = -33.1 \pm 0.05^{\circ}\text{C}}{T_h} = -29.8 \pm 0.03^{\circ}\text{C}$ 

4

4

*h*<sub>(2 *h*)</sub> 19.94 mm

Ti

5

5



**Objective** X Heat transfer coefficient ( $K_v$ ) calculation

Objective

### Heat transfer coefficient $(K_v)$ calculation



Parameters	TVIS
Drying rate at steady state (g/h) (2-2.8 h into primary drying)	0.42
Shelf Temperature, $T_s$ (K)	273.3
Vial's base Temperature, $T_b$ (K)	243.3

Х





 $K_{\nu}(270 \ \mu bar) = 5.73 \times 10^{-4} cal \cdot s^{-1} \cdot cm^{-2} \cdot K^{-1}$ 



## **Additional comments**



Qualification of steady state heat transfer mechanisms



## A single vial technique

*Pikal, et al. (1984)* 



Vol. 73, No. 9, September 1984



Figure 1-Schematic of the laboratory freeze-dryer (see text for key).

The mean sublimation rate was calculated from the mass of ice sublimed and the time required for sublimation.

Product	N	A <sub>v</sub>	$e_v \pm \sigma_m$
H <sub>2</sub> O	7	4.71	0.83 ± 0.04
H <sub>2</sub> O	3	6.83	$0.94 \pm 0.02$
H <sub>2</sub> O	3	17.2	$0.79 \pm 0.03$
KCl (l = 0)	2	4.71	0.88
KCl(l = 0.3)	1	4.71	0.97
KCI(l = 0)	1	20.8	0.58
KCl(l = 0.2)	1	20.8	0.80
Mean			0.84

Table IV-Evaluation of Heat Transfer by Top Radiation: Effective

Emissivity, e.

curred such that ice near the vial wall and ice near the thermocouple wire was preferentially removed. As a result of this phenomenon, measurements of temperature distribution in the ice had to be completed early in the experiment, before the assumption of a planar ice-vapor interface was seriously violated. Accurate temperature distribution data was obtained until  $\sim 15\%$  of the ice had been removed. The vial heat transfer coefficient is defined assuming the ice at the vial bottom is in good thermal contact with the glass. Normally, with vials filled with pure water, partial loss of thermal contact occurs after sub-limation of 35-50% of the ice. Thus, duration of a heat transfer experiment is limited to a time corresponding to sublimation of  $\sim 25\%$  of the ice. Loss of thermal contact is rarely a problem when a frozen solution is dried.

For single vial heat transfer studies, a representative vial from a given lot of vials was modified as shown in Fig. 1. After filling, normally with pure water, the modified vial and other vials of the same lot, all equipped with "identical" metal tubes, were loaded into the laboratory dryer, the liquid was frozen, and the chamber was evacuated. The procedure then involved a series of heat transfer measurements under steady-state conditions at selected shelf temperatures and chamber pressures. An operational definition of steady state is taken as constant temperatures ( $\pm 0.2^{\circ}$ C) and pressures ( $\pm 2 \mu m$ ) for a period of 10-15 min. The sublimation rate,  $\dot{m}$ , is calculated from the observed steady-state pressure readings using Eq. 3 with the closure resistance given by the tube resistance, Eq. 17. The heat transfer rate,  $\dot{Q}$ , is then calculated:

 $\dot{Q}$  (cal/s) = 0.1833 $\dot{m}$  (g/h) (Eq. 18)



## Assumption for $K_v$ determination



- How do we know that the heat transfer mechanisms are constant up to 25% loss of ice mass?
- If the heat transfer mechanisms change because of ice-glass interface contact or because of the change of ice shape (surface area) then surely heat transfer coefficient will change?
- It requires a technique to <u>qualify</u> when the heat transfer mechanisms change
- So can TVIS demonstrate when ice leaves the glass wall interface?



## Limitation of TVIS System ?







## Discussion



- Decrease in  $F_{PEAK}$  suggests that the temperature may be decreasing after the steady state period, contrary to accepted knowledge that the temperature starts to increase owing to a reduction in drying rate and hence the degree of self cooling
- Decrease in  $F_{PEAK}$  is more likely to be due to a change in the ice-glass contact associated with a change in the shape of the ice cylinder.

## Conclusion

- The period for determining the drying rate should be decreased from 25% ice loss to 10% for TVIS to give reliable estimates for Kv
- Opportunity to cycle through shelf temperature and chamber pressure to create the design space for Kv determinations as a function of shelf position.



## Limitations

- $C''_{PEAK}$  and  $F_{PEAK}$  parameters rely on intimate contact of ice cylinder with glass wall
- Cable length limited to 1m at present
- C-TVIS not compatible with front loading system
- Incompatible with TCs in same TVIS vial (use fibre optic sensors INFAP)





## **Future Work**

- Development dryer mapping of sublimation characteristics
  - heat transfer coefficients ( $K_V$ )
  - dry layer resistance ( $R_P$ )





- Instrument Development
  - Contact C-TVIS instrument (2018)
  - Non-contact TVIS (2018-19)
    - Micro-well screening
    - Vial clusters in batch FD
  - TVIS Shuttle (2019-20)





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